

## ***Factsheet:* SI Units**

There are seven basic units used in scientific notation as defined by the **International System of Units (SI)**. The following tables describe these basic units and their use in common physical concepts.

Quantity	SI System Unit		Definition <sup>3</sup>
	Name	Symbol	
Length	meter	m	1 meter = distance travelled by light in 1/299 792 458 s.
Time	second	s	1 second = 9 192 631 770 periods of the radiation equal to the transition between the two hyperfine ground state levels of the <sup>133</sup> Cs atom.
Mass <sup>1</sup>	kilogram	kg	1 kg = mass of the international Pt/Ir prototype of the kilogram.
Electric Current	ampere	A	1 A produces a force of $2 \times 10^{-7}$ N between two, straight, parallel conductors of infinite length and negligible circular cross-section, placed 1 m apart in vacuum.
Thermodynamic Temperature <sup>2</sup>	kelvin	K	1 K = 1/273.16 of the thermodynamic temperature of the triple point of water.
Amount of Substance	mole	mol	1 mol = the amount of C atoms in 0.012 kg <sup>12</sup> C.
Luminous Intensity	candela	cd	1 Cd = the luminous intensity, of a source emitting monochromatic radiation of frequency $540 \times 10^{12}$ Hz in a given direction that has radiant intensity of 1/683 W sr <sup>-1</sup> .

Notes:

1. The SI unit of mass is the kilogram = 1000 grams. No other SI unit has a prefix in its basic form.
2. The SI unit of temperature is the kelvin, not degrees Celsius.
3. <http://physics.nist.gov/cuu/Units>

Other basic concepts have units comprising the seven fundamental SI units.

Quantity	Unit
Area	m <sup>2</sup>
Volume	m <sup>3</sup>
Density	kg m <sup>-3</sup>
Velocity, Speed	m s <sup>-1</sup>
Acceleration	m s <sup>-2</sup>
Wave Number	m <sup>-1</sup>
Molar Mass	kg mol <sup>-1</sup>
Luminance	cd m <sup>-2</sup>

All other quantities have units comprising the basic seven. Take some other well-known units.

Quantity	Unit		SI Unit Form	
	Name	Symbol	Derived	Pure
Force	newton	N	-	$\text{kg m s}^{-2}$
Frequency	hertz	Hz	-	$\text{s}^{-1}$
Electric Charge	coulomb	C	-	A s
Energy, Work	joule	J	N m	$\text{kg m}^2 \text{s}^{-2}$
Pressure	pascal	Pa	$\text{N m}^{-2}$	$\text{kg m}^{-1} \text{s}^{-2}$
Potential Difference	volt	V	$\text{J C}^{-1}$	$\text{m}^2 \cdot \text{kg} \cdot \text{s}^{-3} \cdot \text{A}^{-1}$
Power	watt	W	$\text{J s}^{-1}$	$\text{kg m}^2 \text{s}^{-3}$
Capacitance	farad	F	$\text{C V}^{-1}$	$\text{m}^{-2} \cdot \text{kg}^{-1} \cdot \text{s}^4 \cdot \text{A}^2$
Electrical Resistance	ohm	$\Omega$	$\text{V A}^{-1}$	$\text{m}^2 \cdot \text{kg} \cdot \text{s}^{-3} \cdot \text{A}^{-2}$
Magnetic Flux	weber	Wb	V s	$\text{m}^2 \cdot \text{kg} \cdot \text{s}^{-2} \cdot \text{A}^{-1}$
Magnetic Flux Density	tesla	T	$\text{Wb m}^{-2}$	$\text{kg} \cdot \text{s}^{-2} \cdot \text{A}^{-1}$

## Want to know more?

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